lactone), 1600 (C=C); MS m/z (rel. int.): 162.068 (M⁺, 100) (C₁₀H₁₀O₂), 133 (M - CHO, 18), 104.060 (M - CH₂OCO, 68) (C₈H₈).

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REFERENCES

- Bohlmann, F., Zdero, C., Berger, D., Suwita, A., Mahanta, P. K. and Jeffrey, C. (1979) Phytochemistry 18, 79.
- 2. Bohlmann, F. and Suwita, A. (1967) Chem. Ber. 109, 2021.
- Bohlmann, F., Zdero, C., Jakupovic, J., Robinson, H. and King, R. M. (1981) Phytochemistry 20, 2239.
- Bohlmann, F., Burkhardt, T. and Zdero, C. (1973) Naturally Occurring Acetylenes. Academic Press, London.

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A GUAIANOLIDE FROM CENTAUREA BEHEN*

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Key Word Index—Centaurea behen; Compositae; sesquiterpene lactones; guaianolides.

Abstract—The aerial parts of Centaurea behen afforded in addition to known lactones a new derivative of solstitialin A.

Centaurea behen L., native in Iran, had not been investigated chemically. The aerial parts of this plant afforded several sesquiterpene lactones, the guaianolides cynaropicrin (1) [1], arguerin B (3) [2], desacylcynaropicrin (4) [3], grosshemin (6) [4] and minor amounts of the ketone 8, which is closely related to solstitialin A, the absolute configuration of which had been established [5]. 8 could only be isolated as its diacetate 9, which still was mixed with the acetate of 6. The latter, however, could be

separated from 9 after transformation to the corresponding pyrazoline derivative. The structures of 1, the corresponding diacetate 2, 3, 4 and 6 were elucidated by their 1H NMR data, which are presented in part in Table 1, as good spectra are not available in the literature. The structure of 8 also followed from the 1H NMR data of the corresponding diacetate 9 (Table 1). At 400 MHz in C_6D_6 all signals could be assigned by careful spin decoupling. Starting with the methyl doublet at δ 1.28 H-4

was assigned. H-4 was coupled with a signal at δ 2.12 which was assigned to H-5, as it also was coupled with the proton under the lactone oxygen. H-4 further showed a W-coupling with the three-fold doublet at δ 2.13, which obviously was the H-2 β -signal, as it was further coupled with a proton, which also was coupled with H-5. Finally, by further decoupling all signals could be assigned. The

^{*}Part 352 in the series "Naturally Occurring Terpene Derivatives". For Part 351 see Bohlmann, F., Zdero, C. and Gupta, R. K. (1981) Phytochemistry 20, 2024.

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Table 1. ¹H NMR spectral data of compounds 1-4 and 9 (400 MHz, CDCl₃, TMS as internal standard)

	1	2	3	4	$9(C_6D_6)$
Η-1α	2.98 ddd (br)	3.03 ddd (br)	2.98 ddd (br)	2.95 ddd (br)	2.26 dd (br)
Η-2α	1.74 ddd	1.79 ddd	1.73 ddd	1.73 ddd	1.90 dd
Η-2β	2.24 ddd	2.36 ddd	2.24 ddd	2.23 m	2.13 ddd
Η-3α	$4.56 \ t \ (br)$	$5.56 \ t \ (br)$	$4.56 \ t \ (br)$	4.55 t (br)	
Η-4β			_	_	1.96 ddq
Η-5α	$2.85 \ t \ (br)$	$2.86 \ t \ (br)$	$2.85 \ t \ (br)$	$2.80 \ t \ (br)$	2.12 ddd
Η-6β	4.27 dd	4.16 dd	4.25 dd	4.15 dd	3.71 dd
Η-7α	3.20 dddd	3.20 dddd	3.19 dddd	3.07 dddd	3.03 ddd
H-8a		_	_		1.58 dddd
H-8 <i>β</i>	5.14 ddd	5.16 ddd	5.09 ddd	3.96 ddd	0.96 dddd
Η-9α	2.72 dd	2.69 dd	2.70 dd	2.67 dd	1.44 ddd (br)
Η-9β	2.40 dd	2.41 dd	2.38 dd	2.29 dd	2.04 ddd
H-13	6.24 d	6.24 d	6.22 d	6.25 d	4.38 d
H-13'	5.63 d	5.63 d	5.61 d	6.15 d	4.34 d
H-14	$5.15 \ s \ (br)$	$5.14 \ s \ (br)$	$5.14 \ s \ (br)$	$5.11 \ s \ (br)$	$4.56 \ s \ (br)$
H-14'	$4.94 \ s \ (br)$	$4.96 \ s \ (br)$	$4.94 \ s \ (br)$		$4.34 \ s \ (br)$
H-15 H-15'	5.49 dd 5.37 dd	5.53 dd 5.36 dd	5.50 dd 5.37 dd	5.46 dd } 5.33 dd }	1.28 d
OAc		2.09 s			1.62 s
					1.59 s
OR	$6.34 \ s \ (br)$	$6.56 \ s \ (br)$	$6.19 \ s \ (br)$	_	-
	$5.97 \ s \ (br)$	$5.98 \ s \ (br)$	$5.68 \ s \ (br)$		
	$4.39 \ s \ (br)$	$4.84 \ s \ (br)$	$2.00 \ s \ (br)$	_	

J (Hz): $1\alpha,2\alpha=10.5$; $1\alpha,2\beta=7.5$; $1\alpha,5\beta=9$; $2\alpha,2\beta=13.5$; $2\alpha,3=7.5$; $2\beta,3=7.5$; 3,15=1.5; 3,15'=1.5; 5,6=10.5; 6,7=9; 7,8=9.5; 7,13=3.5; 7,13'=3.0; $8,9\alpha=5.5$; $8,9\beta=4$; $9\alpha,9\beta=14.5$; compound **9**: $1\alpha,2\alpha=9$; $1\alpha,2\beta=1.5$; $1\alpha,5\alpha=9$; $2\alpha,2\beta=19$; $2\beta,4\beta=1.5$; $4\beta,5\alpha=10$; $4\beta,15=7$; $5\alpha,6\beta=6\beta,7\alpha=9$; $7\alpha,8\alpha=2.5$; $7\alpha,8\beta=12$; $8\alpha,8\beta=12$; $8\alpha,9\alpha=5$; $8\alpha,9\beta=2.5$; $8\beta,9\alpha=12$; $8\beta,9\beta=5$; $9\alpha,9\beta=13$; 13,13'=12.

stereochemistry at C-11 was proposed by analogy to that of solstitialin A and by the chemical shift of H-7, which obviously required a *cis*-orientated acetoxy group. 8 therefore is 4β ,15-dihydro-3-dehydro solstitialin A.

EXPERIMENTAL

The fresh plant material (500 g) (collected near Teheran) was extracted with CHCl₃. The polar fractions of the CC (Si gel, Et₂O and Et₂O-MeOH, 20:1) afforded a mixture of 1, 3, 4, 6 and 8, which was further separated by TLC (Si gel). While 1 (100 mg) and 3 (30 mg) could be isolated in a pure state, 4 (20 mg) could be separated from 6 and 8 only by HPLC (reversed phase, RP2, MeOH-H₂O, 7:3). Acetylation of the remaining mixture (in CHCl₃ with Ac₂O-4-pyrrolidinopyridine, room temp., 6 hr) gave the acetates 7 and 9, which again could not be separated

(TLC, Et_2O -petrol, 3:1). Addition of CH_2N_2 , however, afforded the pyrazoline derivative of 7 (2 mg), which could be separated from 9 (2 mg) by TLC (Et_2O -petrol, 3:1).

 4β ,15-Dihydro-3-dehydrosolstitialin A diacetate (9). Colourless gum, IR $v_{\text{max}}^{\text{CCI}_4}$ cm⁻¹: 1805 (γ -lactone), 1755 (C=O, OAc), 1725 (OAc); MS m/z (rel. int.): 364.152 (M⁺, 4) (C₁₀H₂₄O₇), 322 (M - ketene, 2), 304 (M - HOAc, 10), 262 (304 - ketene, 21), 244 (304 - HOAc, 41), 57 (100);

$$[\alpha]_{24}^{\frac{1}{2}} = \frac{589}{+87} \frac{578}{+90} \frac{546}{+105} \frac{436}{+213} \frac{365 \text{ nm}}{+483}$$

$$(c \sim 0.15, \text{ CHCl}_3).$$

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REFERENCES

- Suchy, M., Herout, V. and Šorm, F. (1960) Collect. Czech. Chem. Commun. 25, 2777.
- Gonzales, A. G., Bermejo, J., Cabrera, I., Massanet, G. M., Mansila, H. and Galindo, A. (1971) Phytochemistry 17, 955.
- Gonzales, A. G., Bermejo, J., Massanet, G. M. and Perez, J. (1973) Ann. Quim. 69, 1333.
- Corbella, A. P., Garriboldi, P., Jommi, G., Samek, Z., Holub, M. and Drozdz, B. (1972) Chem. Commun. 386.
- Thiessen, W. E. and Hope, H. (1970) Acta Crystallogr. Sect. B 26, 554.

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AN ISOMER OF XANTHANOL FROM XANTHIUM ORIENTALE*

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(Received 13 January 1981)

Key Word Index—Xanthium orientale; Compositae; Heliantheae; sesquiterpene lactones; xanthanolides.

Abstract—The aerial parts of Xanthium orientale afforded an isomer of xanthanol.

From the genus Xanthium several xanthanolides were isolated [1-11], which seem to be characteristic for this genus, though this type of sesquiterpene lactone has been reported from other genera too. We now have reinvestigated the aerial parts of Xanthium orientale from which the presence of xanthine was reported [7]. Again, only xanthanolides were isolated, the ketones 1 [3] and 2 [2] and the hydroxy acetate 5, which was not identical with xanthanol (3) [1]. The ¹H NMR data of 3 and of the corresponding diacetate 4 were nearly identical with those of the new lactone and the diacetate (Table 1), but there were small characteristic differences in the couplings of H-2 and H-3. Also some of the chemical shifts were slightly different. By spin decoupling, all signals could be assigned. As the couplings of the ring protons were identical with those of 3 and 4, the only possible difference was in the stereochemistry at C-2 or C-4, respectively. The observed shift differences of H-5, however, would favour epimers at C-2. Different stereochemistry at C-10 was unlikely as the couplings of H-10 and the chemical shift of H-14 were nearly identical in 4 and 6. Also a C-8 epimer was not possible, as these epimers characteristically differ in the ¹H NMR spectra [12]. Therefore the new lactone most probably is 2-epixanthanol (5). Unfortunately, the configuration at C-2 is not known for xanthanol and related lactones, which, however, all seem to have the stereochemistry of xanthanol at C-2, if $J_{2,3}$ is conclusive. These results again show that the chemistry of the genus Xanthium is very uniform. This genus is placed in the

Heliantheae, subtribe Ambrosiinae, a very distinctive group [13], mainly characterized by pseudoguaianolides;

Table 1. ¹H NMR spectral data of compounds 4, 5 and 6 (400 MHz, CDCl₃, TMS as internal standard)

	4	5	6
H-2	5.22 dd	5.41 dd	5.19 dd
H-3	2.10 m	1.85 ddd	1.88 ddd
H-3'	1.72 m	1.59 ddd	1.75 m
H-4	4.85 ddq	3.75 ddq	4.91 ddq
H-5	5.87 br dd	5.96 br dd	5.93 br dd
Η-6α	2.53 ddd	2.57 ddd	2.53 ddd
Η-6β	$2.10 \ m$	2.14 ddd	2.13 ddd
H-7	2.52 m	2.48 dddd	2.44 dddd
H-8	4.38 ddd	4.52 ddd	4.28 ddd
Η-9α	1.72 m	1.76 ddd	1.79 m
Н-9β	2.33 ddd	2.35 ddd	2.31 ddd
H-10	2.79 ddq	2.82 ddq	2.79 ddq
H-13	6.17 d	6.20 d	6.17 d
H-13'	5.45 d	5.48 d	5.44 d
H-14	1.11 d	1.18 d	1.10 d
H-15	1.25 d	1.23 d	1.26 d
OAc	2.06 s	2.12 s	2.04 s
	2.04 s		2.03 s

^{*}Part 365 in the series "Naturally Occurring Terpene Derivatives". For Part 364 see Dominguez, X. A., Franco, R., Cano, G., Bapuji, M. and Bohlmann, F. (1981) *Phytochemistry* 20, 2297.

J (Hz): 2,3 = 3; 2,3' = 10.5; 3,3' = 13.5; 3,4 = 10; 3',4 = 3; 4,15 = 6.5; 5,6 α = 9.5; 5,6 β = 3; 6 α ,7 = 2; 6 β ,7 = 10; 7,8 = 10; 7,13 = 3; 8,9 α = 12; 8,9 β = 3; 9 α ,9 β = 12.5; 9 α ,10 = 9 β ,10 = 3.5; 10,14 = 7; (compound 4: 2,3 = 2,3' = 7).